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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/568,832	02/17/2006	Felix Kollmer	HH 307-KFM	4888
10037 7590 01/11/2011 ECKERT SEAMANS CHERIN & MELLOTT, LLC U.S. STEEL TOWER 600 GRANT STREET PITTSBURGH, PA 15219-2788				
EXAMINER JOHNSTON, PHILLIP A				
ART UNIT 2881		PAPER NUMBER		
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**Please find below and/or attached an Office communication concerning this application or proceeding.**

The time period for reply, if any, is set in the attached communication.

# Office Action Summary

**Application No.**

10/568,832

**Applicant(s)**

KOLLMER ET AL.

**Examiner**

PHILLIP A. JOHNSTON

**Art Unit**

2881

**-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --**  
**Period for Reply**

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

**Status**

- 1) ☒ Responsive to communication(s) filed on 02 September 2010.  
2a) ☐ This action is **FINAL**. 2b) ☒ This action is non-final.  
3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

**Disposition of Claims**

- 4) ☒ Claim(s) 1-7 is/are pending in the application.  
4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.  
5) ☐ Claim(s) \_\_\_\_\_ is/are allowed.  
6) ☒ Claim(s) 1-7 is/are rejected.  
7) ☐ Claim(s) \_\_\_\_\_ is/are objected to.  
8) ☐ Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

**Application Papers**

- 9) ☐ The specification is objected to by the Examiner.  
10) ☒ The drawing(s) filed on 12 July 2004 is/are: a) ☒ accepted or b) ☐ objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).  
11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

**Priority under 35 U.S.C. § 119**

- 12) ☒ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).  
a) ☒ All b) ☐ Some \* c) ☐ None of:  
1. ☒ Certified copies of the priority documents have been received.  
2. ☐ Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.  
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

**Attachment(s)**

- 1) ☒ Notice of References Cited (PTO-892)  
2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-940)  
3) ☐ Information Disclosure Statement(s) (PTO/SB/08)  
Paper No(s)/Mail Date \_\_\_\_\_  
4) ☐ Interview Summary (PTO-413)  
Paper No(s)/Mail Date \_\_\_\_\_  
5) ☐ Notice of Informal Patent Application  
6) ☐ Other: \_\_\_\_\_

***Detailed Action***

1. This Office Action is submitted in response to the RCE filed 9-2-2010, wherein claims 1 and 6 have been amended. Claims 1-7 are pending.

***Response to Arguments***

2. Applicant's arguments filed 9-2-2010 have been fully considered but they are not persuasive.

3. No new arguments were submitted in the RCE filed 9-2-2010, which included the following statement; This rejection is respectfully traversed for the reasons previously stated in applicants' Responses to the various prior Office Actions in this case. Therefore the examiner will respond to the previous arguments filed 3-29-2010.

Nowhere in the remarks filed 3-29-2010, did the applicant provide any argument directed against the combination of the Schultz, Orloff and Liebl references. The Applicant's arguments filed 3-29-2010 were directed against each prior art reference individually.

Pages 7-13 are directed against the Schultz reference separately concluding that the Shultz reference only teaches SIMS using an Au primary ion source.

Pages 13 and 14 are directed against the Orloff reference separately and the combination of Orloff and Schulz.

Page 15 is directed solely against the Liebl reference.

In response to applicant's arguments against the references individually, one cannot show nonobviousness by attacking references individually where the rejections are based on combinations of references. See *In re Keller*, 642 F.2d 413, 208 USPQ 871 (CCPA 1981); *In re Merck & Co.*, 800 F.2d 1091, 231 USPQ 375 (Fed. Cir. 1986).

The applicant has failed to present an argument against the rationale for combining the Schultz, Liebl and Orloff references.

4. The Applicant argues at pages 10-12 of the remarks filed 3-29-2010 that, the Schultz reference does not teach the claimed "efficiency of secondary ion production", because it is defined in the instant application as; the secondary ion signal per consumed target material.

In response to this argument the examiner has noted that the feature upon which applicant relies (i.e., the secondary ion signal per consumed target material) is not recited in the rejected claim(s). Although the claims are interpreted in light of the specification, limitations from the specification are not read into the claims. See *In re Van Geuns*, 988 F.2d 1181, 26 USPQ2d 1057 (Fed Cir 1993).

***Claims Rejection – 35 U.S.C. 103***

5. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which the subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

6. Claims 1-7 are rejected under 35 U.S.C. 103(a) as being unpatentable over USPN 6,989,528 to Schultz, in view of Liebl, USPN 3,508,045, and in further view of Orloff, USPN 4,426,582.

7. Regarding claims 1 and 6, Schultz teaches at Col. 5, line 46-67, a secondary ion mass spectrometer (SIMS) apparatus that includes;

(a) A liquid metal ion source (37) for irradiating sample (1) with primary ion beam (4) and creating secondary ion particles (note Figure 7; see Col. 8, line 48-67, and Col. 9, line 50-61),

where a mixed ion beam is initially emitted by the ion source (37) containing metal cluster ions with various charge states and sizes (cluster statuses); for example, gold ions ( $Au_n^{n+}$ ). Col. 5, line 46-63,

(b) A spectrometer unit, time of flight mass spectrometer (22) for detecting the generated secondary ions in a SIMS mode. Col. 8, line 48-67, and Col. 9, line 50-61,

(c) The mixed ion beam is filtered with a Wien filter to provide a mass pure primary ion beam (4) that includes only ions having a specific  $m/z$  at the target surface. See Col. 5, line 46-67 and Col. 9, line 50-61,

(c) Measuring efficiency of secondary ion emission for cluster sizes ranging from a singly charged single gold ion ( $Au^+$ ) to multiply charged clusters having  $n$  atoms ( $Au_n^{n+}$ ). Col. 5, line 46-67.

Regarding increasing secondary ion production efficiency, Schultz measures the increased efficiency or yield of secondary ion production resulting from bombarding the sample with gold (Au) ions of increasing size such as ions ranging in size from  $Au_1^+$  to  $Au_n^{n+}$ . Col. 6, line 13-67.

Schultz discloses a SIMS apparatus that uses a mass pure primary ion beam for bombarding a sample, which allows two and three dimensional depth profiling of large biomolecules, small molecules such as drugs, small inorganic molecules, and elements in biotissues. Col. 5, line 27-67. Schultz also discloses varying the size of the bombarding gold ions in order to increase the secondary ion generation efficiency or yield. Col. 6, line 13-17.

Schultz fails to disclose bombarding the sample with Bismuth ions to increase the efficiency of secondary ion production from the sample, relative to bombardment of the sample with  $Au_1^+$  gold ions.

Liebl discloses at Col. 7, line 44-46 that, in order to generate the maximum number of secondary ions, the mass of the primary ions should be as large as possible.

Liebl modifies the combination of Schultz and Orloff to provide empirical results that show secondary ion yield is directly proportional to mass of the primary ion and is supported by a theory that secondary ion emission is equivalent to the yield of sputtered particles which increases with the atomic mass of the primary ions and with their energies. Col. 7, line 50-70.

One of ordinary skill recognizes that Bismuth (Bi) has a higher atomic number than Gold (Au) and thus for the same charge and cluster state, Bismuth has a higher atomic mass than Gold, which as taught in the prior art Lieble reference would produce a higher secondary ion yield than Gold, when Bismuth is used as the primary ion for bombarding a sample surface and producing a highly sensitive indication of the composition of the material.

Therefore it would have been obvious to one of ordinary skill in the art that Schultz would bombard a sample with a Bismuth primary ion beam since Bismuth has a higher mass than Gold and as predicted by Liebl, would increase the efficiency of secondary ion production from the sample, relative to bombardment of the sample with gold ions, thereby providing increased sensitivity for the analysis of the secondary molecular and elemental ions created during bombardment of a sample surface with a SIMS apparatus. See Liebl; Col. 1, line 32-39 and Schultz; Col. 1, line 14-23 and Col. 5, line 27-31.

Regarding the use of a Bismuth coated liquid metal ion source, the combination of Schultz and Lieble, as described above discloses the use of a liquid metal source in a SIMS apparatus that produces a mass pure primary ion beam for bombarding a sample, where use of a Bismuth primary ion beam for bombarding the surface provides increased sensitivity for the analysis of the secondary molecular and elemental ions. Col. 5, line 27-67.

The combination of Schultz and Lieble fails to explicitly teach using a liquid-metal emitter coated with pure metallic Bismuth or of a low-melting-point alloy containing such that a Bismuth ion mixed beam can be emitted by the ion emitter under the influence of an electric field,

Orloff teaches a liquid metal ion source having emitter 11B, which is coated with liquid metal, such as Bismuth, where the liquid metal attains a very intimate, uniform wetting of the material of the emitter. See Col. 4, line 1-14; Col. 6, line 12-31; and Col. 7, line 62-67.

Orloff modifies the combination of Schultz and Lieble to provide a simple drawn tungsten field emitter coated with Bismuth, with a variable emission current over the nanoamp to microamp range shown in Figures 2A, 2B and 3). See Col. 9, line 53-59.

One of ordinary skill would use a liquid metal ion source that emits Bismuth in order to provide Bismuth ions in the primary ion beam for performing depth profiling with a SIMS apparatus.

Therefore, it would have been obvious to one of ordinary skill that the combination of Schultz and Lieble would use the Bismuth coated emitter of Orloff to provide an ion source for producing high current, medium energy Bismuth ions in a SIMS apparatus, thereby performing

two and three dimensional depth profiling of. See Orloff; Col. 1, line 12-16 and Schultz; Col. 5, line 32-45.

6. Regarding claim 2, the combination of Schultz, Liebl and Orloff discloses a mass-pure primary ion beam of the  $\text{Bi}_n^{P+}$  ion type, described above regarding claims 1 and 6.

7. Regarding claim 3, the combination of Schultz, Liebl and Orloff teaches using a time-of flight, secondary ion mass spectrometer, as described above regarding claims 1 and 6.

8. Regarding claim 4, the combination of Schultz, Liebl and Orloff discloses a primary ion beam having the claimed beam current range, described above regarding claims 1 and 6.

9. Regarding claims 5 and 7, the combination of Schultz, Liebl and Orloff discloses the claimed invention except a liquid metal ion source using a Bi-Pb alloy; however, it would have been obvious to one having ordinary skill in the art at the time the invention was made to use a Bi-Pb alloy, since it have been held to be within the ordinary skill of worker in the art to select a known material on the basis of its suitability for the intended use. One would have been motivated to use a Bi-Pb alloy for the purpose of providing a source of Bismuth metal having a lower melting point and vapor pressure than pure Bismuth.

### ***Conclusion***

10. Any inquiry concerning this communication or earlier communications should be directed to Phillip Johnston whose telephone number is (571) 272-2475. The examiner can normally be reached on Monday-Friday from 7:30 am to 4:30 pm. If attempts to reach the examiner by telephone are unsuccessful, the examiners supervisor Robert Kim can be reached at (571)272-2293. The fax phone number for the organization where the application or proceeding is assigned is 571 273 8300.



Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

/Phillip A Johnston/

Primary Examiner, Art Unit 2881

January 9, 2010

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